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Homonojirimycin analogues and their glucosides from Lobelia sessilifolia and Adenophora spp. (Campanulaceae)

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Abstract

2,6-Dideoxy-7-O-(β-D-glucopyranosyl) 2,6-imino-D-glycero-L-gulo-heptitol (7-O-β-D-glucopyranosyl- α -homonojirimycin, 1) was isolated from the 50% methanol extract of the whole plant of *Lobelia sessilifolia* (Campanulaceae), which was found to potently inhibit rice α -glucosidase. Adenophorae radix, roots of *Adenophora* spp. (Campanulaceae), yielded new homonojirimycin derivatives, adenophorine (2), 1-deoxyadenophorine (3), 5-deoxyadenophorine (4), 1-C-(5-amino-5-deoxy-β-D-galactopyranosyl)butane (β-1-C-butyl-deoxygalactonojirimycin, 5), and the 1-O-β-D-glucosides of 2 (6) and 4 (7), in addition to the recently discovered α -1-C-ethylfagomine (8) and the known 1-deoxymannojirimycin (9) and 2R,5R-bis(hydroxymethyl)-3R,4R-dihydroxypyrrolidine (DMDP, 10). Compound 4 is a potent inhibitor of coffee bean α -galactosidase (IC₅₀ = 6.4 μ M) and a reasonably good inhibitor of bovine liver β -galactosidase (IC₅₀ = 34 μ M). Compound 5 is a very specific and potent inhibitor of coffee bean α -galactosidase (IC₅₀ = 0.71 μ M). The glucosides 1 and 7 were potent inhibitors of various α -glucosidases, with IC₅₀ values ranging from 1 to 0.1 μ M. Furthermore, 1 potently inhibited porcine kidney trehalase (IC₅₀ = 0.013 μ M) but failed to inhibit α -galactosidase, whereas 7 was a potent inhibitor of α -galactosidase (IC₅₀ = 1.7 μ M) without trehalase inhibitory activity. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: Lobelia sessilifolia; Adenophorae radix; Campanulaceae; α-Homonojirimycin derivative; 5-Deoxyadenophorine; 5-Deoxyadenophorine-1-O-β-D-glucoside; α-Glucosidase inhibitor; α-Galactosidase inhibitor

1. Introduction

In 1976, 2R,5R-bis(hydroxymethyl)-3R,4R-dihydroxypyrrolidine (DMDP), the β -D-fructofuranose mimic, was found in leaves of the

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legume *Derris elliptica* [1]. DMDP is now being reported from many disparate species of plants and microorganisms [2], which would indicate that this is a common metabolite. In a search for glycosidase inhibitors from plants, the 50% methanol extracts of some species in the Campanulaceae family showed potent inhibitory activities against rice α -glucosidase. Very recently, we have isolated 6-C-butyl-

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DMDP, α -1-C-ethylfagomine, 1,4-dideoxy-1,4-imino-D-arabinitol (D-AB1), 1-deoxynojirimycin (DNJ), and 1-deoxymannojirimycin (DMJ) from Adenophora triphylla var. japonica [3]. The potent inhibitory activity (IC_{50} = $0.1 \,\mu g/mL$) of the 50% methanol extract of A. triphylla var. japonica against rice α-glucosidase was concluded to be mainly due to the presence of DNJ (IC₅₀ = 0.05 μ M). A 50% methanol extract of Lobelia sessilifolia, after preliminary purification by ion-exchange chromatography with Amberlite IR-120B (H⁺ form) and Dowex 1-X2 (OH- form), also showed potent inhibitory activity (IC₅₀ = 0.1μg/mL) against rice α-glucosidase, while the resin-treated hot-water extract of a traditional Chinese crude drug 'Sya-zin' (Adenophorae radix), which is the roots of Adenophora spp., exhibited good inhibitory activities against rice α-glucosidase ($IC_{50} = 5.3 \mu g/mL$), porcine kidney trehalase ($IC_{50} = 5.4 \mu g/mL$), coffee bean α-galactosidase ($IC_{50} = 4.2 \mu g/mL$), and bovine liver β -galactosidase ($I\dot{C}_{50} = 5.4 \mu g$ / mL). The aqueous methanol extract of Camrotundifolia has previously reported to have a potent inhibitory activity against glucosidases, and this inhibition was shown to be due to a high concentration (up to 2% dry weight in leaves and stems) of DMDP, which is the major alkaloid in all parts of this plant [4]. In this paper, we describe the isolation and characterization of polyhydroxylated alkaloids from L. sessilifolia and Adenopharae radix, their structural determination, and their glycosidase inhibitory activities.

2. Results

Isolation and purification of alkaloids.—The alkaloid fraction was obtained by chromatography of the 50% methanol extract of the aerial parts (370 g) of *L. sessilifolia* on an Amberlite IR-120B (H⁺ form) ion-exchange column. The alkaloid fraction was applied to a Dowex 1-X2 (OH⁻ form) column and eluted with water to give compound 1 (32 mg). On the other hand, the hot-water extract of Adenophorae radix (5 kg) was chromatographed with various ion-exchange resins

to give compounds 2 (18 mg), 3 (25 mg), 4 (3 mg), 5 (13 mg), 6 (28 mg), 7 (18 mg), 8 (6 mg), 9 (63 mg), and 10 (380 mg). The HRFABMS, and ¹H and ¹³C NMR spectral data of 1 were completely in accord with those of 7-O-β-Dglucopyranosyl-α-homonojirimycin 25637) isolated from Aglaonema treubii [5] and Hyacinthus orientalis [6]. MDL 25637 has been chemically synthesized as an oral antidiabetic agent [7]. The HRFABMS and ¹H and ¹³C NMR spectral data of **8** were consistent with those of α -1-C-ethylfagomine isolated from A. triphylla var. japonica [3], and compounds 9 and 10 were identified as DMJ and DMDP, respectively, from their FABMS and NMR data (Scheme 1).

Structural determination of compound 2.— The ¹³C NMR spectral analysis of compound 2 revealed the presence of a single methyl, two methylene, and five methine carbon atoms (Table 1). This result and HRFABMS (m/z)192.1237 $[M + H]^+$) established that the molecular formula was C₈H₁₇NO₄. The complete connectivity of the carbon and hydrogen atoms was defined from extensive decoupling experiments and 2D ¹H-¹³C COSY spectral data. From inspection of the ¹³C NMR chemical shift data in D₂O, the methylene triplet at δ 59.2 (C-1) was assigned to the hydroxymethyl carbon and the methine doublets at δ 74.4, 77.2, and 77.6 to C-3, C-4, C-5 bearing the OH groups, respectively. The relatively high-field methine doublets at δ 56.5 (C-6) and 59.8 (C-2) must be bonded to the nitrogen of the piperidine ring. The large coupling constants ($J_{3,4}$ 10.0 Hz, $J_{4,5}$ 9.0 Hz, $J_{5,6}$ 9.8 Hz) seen in the H-3, H-4, and H-5 indicate an all-trans axial orientation of H-3, H-4, H-5, and H-6, and hence the six-membered ring is in a chair conformation. Irradiation of H-6 at δ 2.65 enhanced the NOE intensity of H-4 and the C-1 hydroxymethyl proton. This indicates the α orientation of the C-1 hydroxymethyl group. Thus, compound 2 was determined to be α -1-deoxy-1-C-methylhomonojirimycin or enantiomer. and was it adenophorine.

Structural determination of compound 3.— The ¹³C NMR spectral data (Table 1) of compound 3 revealed the presence of two methyl, a single methylene, and five methine carbon

$$HO_{A}^{-1}$$
 HO_{A}^{-1} HO_{A}^{-1}

Scheme 1.

atoms. This result and HRFABMS (m/z) $176.1280 \text{ [M + H]}^+$) established that the molecular formula was C₈H₁₇NO₃. From inspection of the ¹³C NMR chemical shift data in D₂O, the relatively high-field methine doublets at δ 53.6 (C-2) and 56.1 (C-6) must be bonded to the nitrogen of the piperidine ring. Decoupling experiments and 2D ¹H-¹³C COSY revealed that the methyl group at δ 14.2 and the ethyl group (δ 11.7 and 26.5) were bonded to C-2 and C-6, respectively. The large coupling constants ($J_{3,4}$ 10.0 Hz, $J_{4,5}$ 8.8 Hz, $J_{5,6}$ 10.0 Hz) and an NOE between H-6 and H-4 or the C-1 methyl proton indicate an all-trans axial orientation of H-3, H-4, H-5, and H-6, and an α orientation of the C-1 methyl group. Thus, compound 3 was determined to be 1-deoxyadenophorine.

Structural determination of compound 4.— The ¹³C NMR spectral data (Table 1) of compound 4 revealed the presence of a single methyl, three methylene, and four methine carbon. This result and HRFABMS (*m/z* 176.1280 [M+H]⁺) established that the molecular formula was C₈H₁₇NO₃. Decoupling experiments and 2D COSY defined a linear CH₂–(CH)₃–CH₂–CH–CH₂–CH₃, showing a linear sequence from C-1H/H' to the C-8

methyl group. From inspection of the 13 C NMR chemical shift data in D_2 O (Table 1), the relatively high-field methine doublets at δ 51.9 (C-6) and 60.3 (C-2) must be bonded to the nitrogen of the piperidine ring, and the methylene triplet at δ 59.4 (C-1) was at-tributed to the hydroxymethyl carbon. From the MS and NMR data described above, com-

Table 1 ¹³C NMR chemical shifts ^a for homonojirimycin analogues and their glucosides

Carbon	2	3	4	5	6	7
C-1	59.2	14.2	59.4	61.8	68.4	68.5
C-2	59.8	53.6	60.3	74.9	58.3	58.7
C-3	74.4	75.6	76.0	77.9	74.2	75.7
C-4	77.2	76.5	71.4	71.8	77.3	71.4
C-5	77.6	78.0	40.2	61.1	77.6	40.0
C-6	56.5	56.1	51.9	64.2	56.7	52.2
C-7	26.8	26.5	30.7	33.5	26.8	30.5
C-8	11.7	11.7	12.4	29.6	11.8	12.4
C-9				25.0		
C-10				16.1		
C-1'					105.9	106.0
C-2'					76.0	76.0
C-3'					78.5	78.5
C-4'					72.5	72.5
C-5'					78.8	78.8
C-6'					63.6	63.6

 $^{^{\}rm a}$ Chemical shifts are expressed in ppm downfield from TSP in ${\rm D_2O}.$

pound 4 was presumed to be the 5-deoxy derivative of adenophorine or its isomer. NOEs between H-3 and H-5ax and between H-6 and H-4 or the C-1 hydroxymethyl proton established that 4 is 5-deoxyadenophorine.

Structural determination of compound 5.— The HRFABMS $(m/z 220.1543 [M + H]^+)$ and ten resonances (Table 1) in the ¹³C NMR spectrum of 5 established that the molecular formula was C₁₀H₂₁NO₄, which was the same as that of 6-C-butyl-DMDP isolated from A. triphylla var. japonica [3]. The complete connectivity of the carbon and hydrogen atoms was defined from extensive decoupling experiments and 2D ¹H-¹³C COSY and HMBC spectral data. These data indicated the methine doublets at δ 61.1 (C-5) and 61.8 (C-1) were bonded to the nitrogen of the piperidine ring, bearing the hydroxymethyl (δ 64.2) and butyl groups (δ 16.1, 25.0, 29.6, 33.5), respectively. The strong NOE between H-1 and H-3 or H-5 indicates that these three protons are 1,3-trans-diaxial. The coupling patterns of H-2 $(\delta 3.40, dd, J_{1.2} 9.2 Hz, J_{2.3} 9.9 Hz)$ and H-3 $(\delta$ 3.50, dd, $J_{2,3}$ 9.9 Hz, $J_{3,4}$ 3.3 Hz) imply that H-2, H-3 and H-4 are axial, axial, and equatorial, respectively. Thus, compound 5 was determined to be 'β-1-C-butyl-deoxygalactonojirimycin' or its enantiomer. Very recently, Krülle et al. have enantiospecifically syntheβ-1-C-butyl-deoxygalactonojirimycin sized from D-galacto-1,4-lactone [10]. From the NMR data and specific rotation values (natural sample: $[\alpha]_D - 12.8^\circ$ (c 0.36, water); synthetic sample: $[\alpha]_D - 14.8^\circ$ (c 0.39, water)), both compounds were determined to be identical.

Structural determination of compound 6.— The HRFABMS (m/z 354.1758 [M + H]⁺) and 14 resonances (Table 1) in the ¹³C NMR spectrum of 6 established that the molecular formula was $C_{14}H_{27}NO_9$. The response to the naphthoresorcinol–sulfuric acid reagent and the characteristic anomeric proton (H-1', δ 4.48, $J_{1',2'}$ 8.1 Hz) and carbon (C-1', δ 105.9) signals in the NMR suggested that 6 was a glycoside of an alkaloid. After acid hydrolysis of this glycoside using Dowex 50W-X2 (H⁺) resin, D-glucose was detected in the filtrate by the D-glucose oxidase peroxidase method, while the aglycone part was displaced from

the resin with 0.5 M NH₄OH, concentrated to dryness, and confirmed as adenophorine by direct comparison of its ¹³C NMR spectrum with that of 2. The complete connectivity of the carbon and hydrogen atoms was defined from extensive decoupling experiments and 2D ¹H-¹³C COSY spectral data. From the coupling constant of the anomeric proton of this D-glucoside, the type of glucosidic linkage was determined to be β. In the ¹³C NMR spectrum, glucoside formation for adenophorine produced a 9.2-ppm downfield shift for C-1 and a 1.5-ppm upfield shift for C-2, while the chemical shifts of C-3 to C-8 remained unchanged (Table 1). Therefore, the structure of 6 was shown to be 1-O-β-Dglucopyranosyladenophorine.

Structural determination of compound 7.— The HRFABMS $(m/z \ 338.1817 \ [M + H]^+)$ and 14 resonances (Table 1) in the ¹³C NMR spectrum of 7 established that the molecular formula was C₁₄H₂₇NO₈. From the response to the naphthoresorcinol-sulfuric acid reagent and the characteristic anomeric proton (H-1', δ 4.47, $J_{1',2'}$ 8.1 Hz) and carbon (C-1', δ 106.0) signals in the NMR, compound 7 was also found to be a glycoside of an alkaloid. Acid hydrolysis of 7 in the same manner as described above gave D-glucose and 5-deoxyadenophorine. The linkage site was determined from the observation of strong NOEs from the anomeric proton to the C-1 hydroxymethyl protons (H-1a and H-1b) of the aglycone, obtained from 2D NOESY. Thus, compound 7 was determined to be 1-Oβ-D-glucopyranosyl-5-deoxyadenophorine.

Glycosidase inhibitory activities of the alkaloids.—The IC₅₀ values of homonojirimycin analogues and their glucosides against various glycosidases are shown in Table 2. We have reported that α-homonojirimycin (HNJ) is a potent inhibitor of rice α-glucosidase and all types of mammalian α-glucosidases [8]. Adenophorine (2), which is the 1-deoxy-1-C-methyl derivative of HNJ, was approximately a 100-1000-fold weaker inhibitor of α-glucosidases than HNJ, but it was a reasonably good inhibitor of coffee bean α-galactosidase (IC₅₀ = 11μ M). Deoxygenation at C-1 of 2 to give 3 retained inhibitory potential toward α-glucosidases and α-galactosidase, while de-

Table 2 Concentration of homonojirimycin analogues and their glucosides giving 50% inhibition of glycosidase activities

Enzyme	IC_{50} (μ M)										
	1	2	3	4	5	6	7				
α-Glucosidase											
Rice	0.25	34	21	NI	NI	610	0.49				
Rat intestinal sucrase	3.3	20	110	NI	NI	1000	2.4				
Rat intestinal maltase	2.3	150	280	NI	NI	NI	6.1				
Rat intestinal isomaltase	0.27	330	110	NI	NI	880	2.1				
β-Glucosidase											
Almond	NI ^a	NI	NI	NI	NI	NI	NI				
Trehalase											
Porcine kidney	0.013	NI	NI	1000	NI	53	NI				
α-Galactosidase											
Coffee bean	NI	11	23	6.4	0.71	NI	1.7				
β-Galactosidase											
Bovine liver	NI	NI	NI	34	NI	NI	NI				

^a No inhibition (<50% inhibition at 1000 μ M).

oxygenation at C-5 of 2 to give 4 enhanced its inhibitory activity toward α-galactosidase $(IC_{50} = 6.4 \mu M)$ and further generated an inhibitory activity toward β-galactosidase $(IC_{50} = 34 \mu M)$, but abolished its inhibition of α-glucosidases. Although 1-deoxygalactonojirimycin is known as an extremely powerful inhibitor ($K_i = 0.0016 \mu M$) of coffee bean α galactosidase [9], its β -1-C-butyl derivative (5) was also a very specific inhibitor of this enzyme, with an IC_{50} value of 0.71 μM. β-1-C-Butyl-deoxygalactonojirimycin chemically synthesized [10] and its IC₅₀ value toward coffee bean α-galactosidase was 0.69 μM. Thus, the natural compound 5 was identical with the synthetic sample in both the physicochemical aspects and biological activ-7-O-β-D-Glucopyranosyl-α-homonojirimycin (MDL 25637) (1) is a potent inhibitor of α -glucosidases and trehalase [7,11]. 7-0- β -D-Glucopyranosyladenophorine (6), which can be regarded as the 1-deoxy-1-C-methyl derivative of 1, significantly reduced inhibitory potential toward α-glucosidases and trehalase. Very interestingly, the deoxygenation at C-5 of 6 to give 7 remarkably enhanced its inhibitory activity toward α-glucosidases and α-galactosidase.

3. Discussion

Some of the polyhydroxylated alkaloids, such as DMDP, are being reported from

many disparate species of plants, and more recently from microorganisms [2], which would indicate that this is clearly a common metabolite. Others appear to be restricted in distribution. For instance, until 1995 αhomonojirimycin had only been found in the neotropical liana Omphalea diandra (Euphorbiaceae) [12]. However, our recent work and this study elucidated that α -homonojirimycin or its glucoside also occurs in species of the Araceae (A. treubii [5] and Nephthytis poissoni [2]), H. orientalis (Hyacinthaceae) [6], and L. sessilifolia (Lobeliaceae). Similarly, until recently the polyhydroxylated pyrrolizidine alkaloids such as alexine had only been found in two closely related legume genera, Alexa and Castanospermum [2], but since 1994 they have reported from families the acinthaceae [13], Casuarinaceae [14], and Myrtaceae [15]. So, it is probably too early to discuss the significance of the distribution of polyhydroxylated alkaloids until broader surveys have been conducted.

We recently isolated the novel polyhydroxy-lated piperidine alkaloid with a linear eight-carbon backbone, α-1-*C*-ethylfagomine (8), from *A. triphylla* var. *japonica* [3]. Interestingly, compound 8 has the same carbon backbone structure as 5-deoxyadenophorine (4). Compounds 4 and 8 could be synthesized from a common precursor in the biosynthetic pathway. Also 6-*C*-butyl-DMDP from *A. tri-phylla* var. *japonica* [3] and β-1-*C*-butyl-deoxy-

galactonojirimycin (5) from Adenophorae radix could both be biosynthesized by the fiveor six-membered ring closure from a common precursor.

The resin-treated extract of Adenophora radix exhibited a good inhibitory activity against rice α-glucosidase, coffee bean α-galactosidase, bovine liver β-galactosidase, and porcine kidney trehalase. This α-glucosidase inhibitory activity is concluded to be due to a combination of effects of 2 (IC₅₀ = 34 μ M), 3 $(IC_{50} = 21 \mu M)$, 6 $(IC_{50} = 610 \mu M)$, 7 $(IC_{50} =$ 0.49 μ M), 8 (IC₅₀ = 490 μ M), and 10 (IC₅₀ = 300 μ M) on rice α -glucosidase. The α - and β-galactosidase inhibitory activities are attributed to a combination of effects of 2 $(IC_{50} = 11 \mu M)$, 3 $(IC_{50} = 23 \mu M)$, 4 $(IC_{50} =$ $6.4 \,\mu\text{M}$), 5 (IC₅₀ = 0.71 $\,\mu\text{M}$), and 7 (IC₅₀ = 1.7 μM) on coffee bean α-galactosidase and that of 4 (IC₅₀ = 34 μ M), 8 (IC₅₀ = 29 μ M), and 10 $(IC_{50} = 2.2 \mu M)$ on bovine liver β -galactosidase, respectively. With regard to the trehalase inhibition by the resin-treated extract, it would not be sufficient to explain this inhibitory activity only by that of DMDP $(IC_{50} = 500 \mu M)$ and 6 $(IC_{50} = 53 \mu M)$. We previously proved the presence of the two subsites, one for catalysis and one for recognition, on the active center of pig kidney trehalase from a kinetic analysis of the multiple inhibition by two competitive-type monosaccharide analogue inhibitors [16]. DMDP and DMJ are competitive inhibitors, with K_i values of 150 and 390 µM, respectively, and are classified into catalytic-site and recognitionsite targeting inhibitors (Fig. 1). An equimolar mixture of DMDP and DMJ may produce the synergistic interactions with the two subsites, that is, the potent inhibition by the extract may be due to the synergistic effect obtained by multiplying the inhibitory activity for DMDP on the catalytic site times that for DMJ on the recognition site (data not shown).

4. Experimental

General methods.—The purity of samples was checked by HPTLC on Silica Gel 60F₂₅₄ (E. Merck) using the solvent system 4:1:1 PrOH–AcOH–water, and a chlorine–o-to-

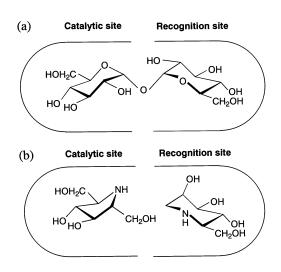


Fig. 1. (a) Trehalose and (b) DMDP and DMJ in the two subsites on the active center of trehalase.

lidine reagent was used for detection. Optical rotations were measured with a Jasco DIP-370 digital polarimeter. ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) spectra were recorded on a Jeol JNM-GX 400 spectrometer. Chemical shifts are expressed in ppm downfield from sodium 3-(trimethylsilyl)propionate (TSP) in D₂O. Mass spectra were measured on a Jeol JMS-SX 102A spectrometer.

Extraction and isolation.—A 50% MeOH extract of the fresh aerial parts (370 g) of L. sessilifolia was applied to a column of Amberlite IR-120B (200 mL, H⁺ form). The 0.5 M NH₄OH eluate was concd to give a brown oil (650 mg), which was chromatographed over Dowex 1-X2 (1.5 \times 63 cm, OH⁻ form) with water as eluant (fraction size 10 mL). Concentration of Fractions 18–23 gave compound 1 (32 mg). A hot-water extract of Adenophorae radix (5 kg) was applied to a column of Amberlite IR-120B (1 L, H⁺ form) and eluted with 0.5 M NH₄OH. The NH₄OH eluate was concd to give a brown oil (59 g), which was applied to a column of Dowex 1-X2 (300 mL, OH⁻ form) and eluted with water (3 L). The water eluate was concd to give a colorless oil (37 g). This oil was chromatographed over a Dowex 1-X2 column (2.5 \times 95 cm, OH⁻form) with water as eluant (fraction size 20 mL). The water eluate was divided into two pools A (Fractions 23–45, 25 g) and B (Fractions 46– 60, 0.85 g). Each pool was further chro-CG-50 matographed on an Amberlite $(2.5 \times 95 \text{ cm}, \text{NH}_{4}^{+} \text{ form})$ column with water as eluant to give **5** (13 mg), **2** (18 mg), **3** (25 mg), **6** (28 mg), **7** (18 mg), **8** (6 mg), **4** (3 mg), and DMJ (63 mg) from pool A in order of elution, and DMDP (380 mg) from pool B.

Glycosidase inhibitory activities.—The enzymes α -glucosidase (from rice), β -glucosidase (from almond), β -galactosidase (from bovine liver), trehalase (from porcine kidney), α -galactosidase (from coffee bean), p-nitrophenyl glycosides, and disaccharides were purchased from Sigma Chemical Co. Brush border membranes, prepared from the intestine of male Wister rats by the method of Kessler et al. [17], were used as the source of rat intestinal glycosidases.

The activities of rice α-glucosidase, rat intestinal glycosidases, and trehalase were determined using the appropriate disaccharides as substrates at the optimum pH of each enzyme. The released D-glucose was determined colorimetrically using the Glucose B-test (Wako Pure Chemical Industries). Other glycosidase activities were determined using an appropriate *p*-nitrophenyl glycoside as a substrate at the optimum pH of each enzyme. The reaction was stopped by adding 400 mM Na₂CO₃. The released *p*-nitrophenol was measured specrometrically at 400 nm.

Adenophorine (2).—[α]_D + 59.7° (c 1.0, water); HRFABMS: m/z 192.1237 [M + H]⁺ (C₈H₁₈NO₄ requires 192.1236); ¹H NMR (400 MHz, D₂O) δ 0.94 (t, 3 H, J 7.6 Hz, CH₃), 1.35 (m, 1 H, H-7a), 1.84 (m, 1 H, H-7b), 2.65 (ddd, 1 H, J 3.4, 8.0, 9.8 Hz, H-6), 3.11 (dd, 1 H, J 9.0, 9.8 Hz, H-5), 3.26 (m, 1 H, H-2), 3.46 (dd, 1 H, J 9.0, 10.0 Hz, H-4), 3.75 (dd, 1 H, J 6.1, 10.0 Hz, H-3), 3.76–3.90 (2 H, H-1a, H-1b); ¹³C NMR data, see Table 1.

1-Deoxyadenophorine (3).—[α]_D + 67.8° (c 0.83, water); HRFABMS: m/z 176.1280 [M + H]⁺ (C₈H₁₈NO₃ requires 176.1287); ¹H NMR (400 MHz, D₂O) δ 0.92 (t, 3 H, J 7.6 Hz, CH₃), 1.13 (d, 3 H, J 7.1 Hz, CH₃), 1.36 (m, 1 H, H-7a), 1.80 (m, 1 H, H-7b), 2.68 (ddd, 1 H, J 3.4, 7.6, 10.0 Hz, H-6), 3.09 (dd, 1 H, J 8.8, 10.0 Hz, H-5), 3.28 (m, 1 H, H-2), 3.53 (dd, 1 H, J 8.8, 10.0 Hz, H-4), 3.60 (dd, 1 H, J 5.6, 10.0 Hz, H-3); ¹³C NMR data, see Table 1. 5-Deoxyadenophorine (4).—[α]_D + 50.1° (c 0.32, water); HRFABMS: m/z 176.1285 [M + H]⁺ (C₈H₁₈NO₃ requires 176.1287); ¹H NMR

(400 MHz, D₂O) δ 0.91 (t, 3 H, J 7.6 Hz, CH₃), 1.16 (m, 1 H, H-5ax), 1.45 (m, 2 H, H-7a, H-7b), 2.09 (m, 1 H, H-5eq), 2.88 (m, 1 H, H-6), 3.32 (m, 1 H, H-2), 3.69–3.74 (2 H, H-3, H-4), 3.75–3.85 (2 H, H-1a, H-1b); ¹³C NMR data, see Table 1.

1-C-(5-amino-5-deoxy-β-D-galactopyranosyl)-butane (β-1-C-butyl-deoxygalactonojirimycin, **5**).—[α]_D — 12.8° (c 0.36, water); HR-FABMS: m/z 220.1543 [M + H]⁺ (C₁₀H₂₂NO₄ requires 220.1549); ¹H NMR (400 MHz, D₂O) δ 0.91 (t, 3 H, CH₃), 1.25–1.47 (5 H, H-7a, H-8a, H-8b, H-9a, H-9b), 1.80 (m, 1 H, H-7b), 2.47 (ddd, 1 H, J 3.3, 7.7, 9.2 Hz, H-1), 2.80 (ddd, 1 H, J 1.5, 6.6, 7.0 Hz, H-5), 3.40 (dd, 1 H, J 9.2, 9.9 Hz, H-2), 3.50 (dd, 1 H, J 3.3, 9.9 Hz, H-3), 3.63 (dd, 1 H, J 6.6, 11.0 Hz, H-6a), 3.67 (dd, 1 H, J 7.0, 11.0 Hz, H-6b), 3.99 (dd, 1 H, J 1.5, 3.3 Hz, H-4); ¹³C NMR data, see Table 1.

1-O- β -D-Glucopyranosyladenophorine (6).— $[\alpha]_D + 24.9^{\circ}$ (c 0.92, water); HRFABMS: m/z $[M + H]^{+}$ (C₁₄H₂₈NO₉ requires 354.1758 354.1764); ¹H NMR (400 MHz, D_2O) δ 0.94 (t, 3 H, J 7.6 Hz, CH₃), 1.18 (m, 1 H, H-7a), 1.83 (m, 1 H, H-7a), 2.71 (ddd, 1 H, J 3.2, 7.8, 9.8 Hz, H-6), 3.12 (dd, 1 H, J 9.3, 9.8 Hz, H-5), 3.31 (dd, 1 H, J 8.1, 9.3 Hz, H-2'), 3.39 (dd, 1 H, J 9.0, 9.8 Hz, H-4'), 3.41–3.50 (3 H, H-2, H-4, H-5'), 3.51 (dd, 1 H, J 9.0, 9.3 Hz, H-3'), 3.73 (dd, 1 H, J 5.9, 12.5 Hz, H-6'a), 3.74 (dd, 1 H, J 6.1, 10.0 Hz, H-3), 3.92 (t, 1 H, J 10.7 Hz, H-1a), 3.92 (dd, 1 H, J 2.2, 12.5 Hz, H-6'b), 4.11 (dd, 1 H, J 3.7, 10.7 Hz, H-1b), 4.48 (d, 1 H, J 8.1 Hz, H-1'); ¹³C NMR data, see Table 1.

 $1 - O - \beta - D - Glucopyranosyl - 5 - deoxyadeno$ phorine (7).— $[\alpha]_D$ + 11.4° (c 0.33, water); $[M + H]^{+}$ m/z**HRFABMS**: 338.1817 $(C_{14}H_{28}NO_8 \text{ requires } 338.1815); {}^1H \text{ NMR } (400)$ MHz, D₂O) δ 0.91 (t, 3 H, J 7.6 Hz, CH₃), 1.17 (m, 1 H, H-5ax), 1.46 (m, 2 H, H-7a, H-7b), 2.10 (ddd, 1 H, J 2.9, 4.4, 12.9 Hz, H-5eq), 2.88 (m, 1 H, H-6), 3.31 (dd, 1 H, J 8.1, 9.3 Hz, H-2'), 3.40 (dd, 1 H, J 9.0, 9.8 Hz, H-4'), 3.47 (ddd, 1 H, J 2.2, 5.9, 9.8 Hz, H-5'), 3.49–3.54 (m, 1 H, H-2), 3.51 (dd, 1 H, J 9.0, 9.3 Hz, H-3'), 3.68–3.73 (m, 2 H, H-3, H-4), 3.73 (dd, 1 H, J 5.9, 12.4 Hz, H-6'a), 3.92 (dd, 1 H, J 12.4 Hz, H-6'b), 3.92 (t, 1 H, J 10.7 Hz, H-1a), 4.12 (dd, 1 H, J 3.7, 10.7 Hz, H-1b), 4.47 (d, 1 H, J 8.1 Hz, H-1'); ¹³C NMR data, see Table 1.

Acid hydrolysis of compounds 6 and 7.—A small amount (6 mg) of compound 6 or 7 was heated at 100° C with Dowex 50W-X2 (0.5 g, H⁺ form) in water for 8 h. The resin was filtered off and packed into a short column. The alkaloid moiety was eluted with 0.5 M NH₄OH, concd and lyophilized. On the other hand, after neutalizing the filtrate, the amount of the released D-glucose was determined by the D-glucose oxidase peroxidase method using the commercially available Glucose B-test (Wako).

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